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1. Product and Company Identification

Product Code: FGAU106-109
Product Name: SETT ENHANCED
Trade Name: SETT ENHANCED
Company Name: Stoller Australia Pty Ltd

1 Creswell Road

Largs Bay

South Australia 5016

Web site address: www.stoller.com.au stoller@stoller.com.au

Emergency Contact: STOLLER PRODUCTION CHEMIST

Contact number: 08 8169-0988

Information: 1800 337-845

Intended Use: For agricultural use only

2. Hazards Identification

Serious Eye Damage/Eye Irritation, Category 2

Acute Toxicity: Skin, Category 4
Acute Toxicity: Oral, Category 5



GHS Signal Word: Warning

GHS Hazard Phrases: H303 - May be harmful if swallowed.

H312 - Harmful in contact with skin. H319 - Causes serious eye irritation.

GHS Precaution Phrases: P264 - Wash hands thoroughly after handling.

P280 - Wear protective gloves/protective clothing/eye protection/face protection.

GHS Response Phrases: P302+352 - IF ON SKIN: Wash with plenty of soap and water.

P305+351+338 - IF IN EYES: Rinse cautiously with water for several minutes. Remove

contact lenses, if present and easy to do. Continue rinsing.

P312 - Call a POISON CENTER or doctor/physician if you feel unwell.

P321 - Specific treatment see ... on this label.

P337+313 - If eye irritation persists, get medical advice/attention. P362+364 - Take off contaminated clothing and wash it before reuse.

GHS Storage and Disposal

Phrases:

P501 - Dispose of contents/container to ...

Potential Health Effects Acute: Depending on the duration of contact, overexposure can irritate the eyes, skin,

(Acute and Chronic): mucous membranes and any other exposed tissue.

Inhaling mist, spray, or vapor may cause irritation to upper respiratory tract (nose and

throat).

Skin Contact: Skin irritation. Skin exposure may cause slight irritation, redness, itching, swelling. May

cause more severe response if skin is damp, abraded (scratched or cut), or covered by clothing, gloves or footwear. Prolonged contact may cause more severe symptoms.

Damage is localized to contact areas.

Eye Contact: Causes eye irritation. Eye exposure may cause serious eye irritation, pain, and/or

damage to the eye. May cause conjuntival swelling and cornea opacification from hypertonic solution. Corneal eye pain, redness, acute corneal thickening or whitening.

Ingestion: Consumption of hypertonic solutions causes nausea, vomiting, and increased thirst.

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SAFETY DATA SHEET

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Medical Conditions Generally Any skin condition that disrupts the skin, such as abrasions, cuts, psoriasis, fungal

Aggravated By Exposure: infections, etc. Any eye condition that compromises tear production, conjunctiva, or

normal corneal homestasis.

3. Composition/Information on Ingredients

CAS#	Components (Chemical Name)	Concentration	
22691-02-7	Calcium chloride (CaCl2), hydrate	<20.0 %	
10043-35-3	Boric acid	<15.0 %	
57-50-1	Sucrose	< 5.0 %	

4. First Aid Measures

Emergency and First Aid

Procedures:

Victims of severe exposure to chemicals must be taken to health providing centers for medical attention. Always bring with victim a copy of label and SDS of product to health

professional.

In Case of Inhalation: Remove person to fresh air and keep comfortable for breathing. Call a POISON

CENTER or doctor/physician if you feel unwell.

In Case of Skin Contact: Wash with plenty of water. If skin irritation occurs, get medical advice/attention. Take off

contaminated clothing and wash before re-use. Seek medical attention should severe

irritation occurs.

In Case of Eye Contact: Rinse cautiously with water for several minutes. Remove contact lenses, if present and

easy to do. Continue rinsing. If irritation occurs, get medical advice/attention.

In Case of Ingestion: Rinse mouth. Call a POISON CENTER or doctor/physician if you feel unwell.

Signs and Symptoms Of

Exposure:

Localized redness, warmth, and irritation consistent with mechanism of injury: abrasion,

burn, hypertonic solution.

Note to Physician: Treat symptomatically and supportively.

5. Fire Fighting Measures

Flash Pt: N.A.

Explosive Limits: LEL: N.A. UEL: N.A.

Autoignition Pt: N.A.

Suitable Extinguishing Media: Use extinguishing measures that are appropriate to local circumstances and the

surrounding environment.

Unsuitable Extinguishing

Media:

None known.

Fire Fighting Instructions: Keep unnecessary people away; isolate hazard area and deny entry. Fight fire for other

material that is burning. Water should be applied in large quantities as fine spray. Wear NIOSH approved positive-pressure self-contained breathing apparatus operated in pressure demand mode. Wear protective fire fighting clothing (includes fire fighting helmet, coat, trousers, boots, and gloves). Avoid contact with this material during fire fighting operations. If contact is likely, change to full chemical resistant fire fighting clothing with self-contained breathing apparatus. If this is not available, wear full chemical resistant clothing wih self-contained breathing apparatus and fight fire from a remote location. For protective equipment in post-fire or non-fire clean-up situations,

refer to the relevant sections.

Flammable Properties and

Hazards:

This material does not burn.

Hazardous Combustion

Formed under fire conditions: hydrogen chloride gas, calcium oxide

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6. Accidental Release Measures

Protective Precautions, Protective Equipment and Emergency Procedures: Isolate the area. Keep unnecessary and unprotected personnel from entering the area. Spilled material may cause a slipping hazard on some surfaces. Use appropriate safety equipment. For additional information, refer to Section 8, Exposure Controls and Personal Protection. Refer to Section 7, Handling, for additional precautionary measures.

Prevent entry into waterways, sewers, basements or confined areas. See Section 12,

Ecological Information.

Steps To Be Taken In Case Material Is Released Or Spilled:

Environmental Precautions:

Small and large spills: Contain spilled material if possible. Absorb with materials such as sand. Collect in suitable and properly labeled containers. Flush residue with water. See Section 13, Disposal Considerations, for additional information.

7. Handling and Storage

Precautions To Be Taken in Handling:

Avoid contact with eyes, skin, and clothing. Do not swallow. Wash thoroughly after handling. Wear personal protective equipment as described in Section 8, Exposure Controls/Personal Protection.

Precautions To Be Taken in Storing:

Protect from atmospheric moisture. Keep containers tightly closed when not in use. Keep separated from incompatible substances see Section 10, Stablility and Reactivity.

8. Exposure Controls/Personal Protection

CAS#	Partial Chemical Name	OSHA TWA	ACGIH TWA	Other Limits	
22691-02-7	Calcium chloride (CaCl2), hydrate	No data.	TLV: 10 mg/m³	No data.	
10043-35-3	Boric acid	No data.	TLV: 2 mg/m³ STEL: 6 mg/m³	No data.	
57-50-1	Sucrose	No data.	No data.	No data.	

Respiratory Equipment (Specify Type):

Respiratory protection should be worn when there is a potential to exceed the exposure limit requirements or guidelines. If there are no applicable exposure limit requirements or guidelines, wear respiratory protection when adverse effects, such as respiratory irritation or discomfort have been experienced, or where indicated by your risk assessment process. In dusty or misty atmospheres, use an approved particulate respirator. The following should be effective types of air-purifying respirators: high efficiency particulate air (HEPA) N95. A respiratory protection program that meets OSHA's 29 CFR 1910.134 and ANSI Z88.2 requirements or European Standard EN 149 must be followed whenever workplace conditions warrant respirator use.

Eye Protection:

Wear safety glasses with side-shields. Wear chemical safety goggles and/or a

face-shield to protect against skin and eye contact when appropriate.

Protective Gloves: Use gloves chemically resistant to this material. If hands are cut or scratched, use gloves

chemically resistant to this material even for brief exposures. Handle with gloves. Dispose of contaminated gloves after use in accordance with applicable laws and good

laboratory practices. Wash and dry hands.

Other Protective Clothing: Wear clean, body-covering clothing. Wear appropriate clothing to avoid skin contact.

Wear long sleeve shirt, long pants, impervious gloves and protective shoes.

Engineering Controls (Ventilation etc.):

Use local exhaust ventilation, or other engineering controls to maintain airborne levels below exposure limit requirements or guidelines. If there are no applicable exposure limit requirements or guidelines, general ventilation should be sufficient for most operations.

Local exhaust ventilation may be necessary for some operations.

Work/Hygienic/Maintenance

Practices:

Use good personal hygiene. Do not consume or store food in the work area. Wash hands and affected skin immediately after handling, before smoking or eating, before

breaks, and at the end of the workday.

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	9. Physical and Chemical Properties
Physical States:	[] Gas [X] Liquid [] Solid
Appearance and Odor:	Brown liquid.
рН:	~ 3.5 - 5.5
Freezing Point:	N.E.
Boiling Point:	> 0.00 F
Flash Pt:	N.A.
Evaporation Rate:	N.E.
Flammability (solid, gas):	Material will not burn.
Explosive Limits:	LEL: N.A. UEL: N.A.
Vapor Pressure (vs. Air or mm Hg):	N.E.
Vapor Density (vs. Air = 1):	N.E.
Specific Gravity (Water = 1):	~ 1.32 - 1.36
Density:	~ 11.2 LB/GA
Solubility in Water:	Soluble
Saturated Vapor	N.E.
Concentration:	
Octanol/Water Partition Coefficient:	No data.
Percent Volatile:	N.A.
Autoignition Pt:	N.A.
Decomposition Temperature:	N.A.
Viscosity:	N.E.
	10. Stability and Reactivity
Reactivity:	Hygroscopic.
Stability:	Unstable [] Stable [X]
Conditions To Avoid - Instability:	Stable under normal temperatures and pressures.
Avoid:	Avoid contact with: bromide trifluoride. 2-furan percarboxylic acid because calcium chloride is incompatible with those substances. Contact with zinc forms flammable hydrogen gas, which can be explosive. Catalizes exothermic polymerization of methyl vinyl ether. Reaction of bromide impurity with oxidizing materials may generate trace levels of impurities such as bromates.
Hazardous Decomposition or Byproducts:	Formed under fire conditions: hydrogen chloride gas, calcium oxide
Possibility of Hazardous Reactions:	Will occur [] Will not occur [X]
Conditions To Avoid - Hazardous Reactions:	None known.

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11. Toxicological Information

Toxicological Information: Mutagenicity: This product has not been investigated for mutagenic effects.

Embryotoxicity: This product has not been investigated for embryotoxic effects.

Teratogenicity: This product has not been investigated for teratogenic effects.

Reproductive Toxicity: This product has not been investigated for toxic reproductive

effects.

CAS# 10043-35-3: Acute toxicity, LD50, Oral, Rat, 2660. MG/KG. Result:

Gastrointestinal:Hypermotility, diarrhea. Gastrointestinal:Nausea or vomiting. ; Journal of the American Medical Association, American Medical Association, 535 N. Dearborn

St., Chicago, IL 60610, Vol/p/yr: 128,266, 1945

Irritation or Corrosion: Symptoms related to

No data available.

No data available.

Toxicological Characteristics:

TTO Gata available

Sensitization:

No data available.

Chronic Toxicological Effects:

The toxicological properties of this material have not been fully investigated.

Carcinogenicity/Other

No component is listed as a carcinogenic by IARC, NTP, OSHA, and ACGIH.

Information:

Carcinogenicity:

NTP? No IARC Monographs? No OSHA Regulated? No

12. Ecological Information

General Ecological Information:

The available data on this material does not indicate any undue hazard to the environment under anticipated use and storage. All work practices must be aimed at eliminating environmental contamination. Any waste due to spillage or leakage should be contained and disposed of accordingly, see above under Section 6 "Accidental Release

Measures."

Results of PBT and vPvB

assessment:

No data available.

Persistence and Degradability: Calcium chloride is believed not to persist in the environment because it is readily dissociated into calcium and chloride ions in water. Both ions originally exist in nature, and their concentrations in surface water will depend on various factors, such as

geological parameters, weathering and human activities.

Bioaccumulative Potential: Calcium chloride and its dissociated forms (calcium and chloride ions) are ubiquitous in

the environment. Calcium and chloride ions can also be found as constituents in organisms. Considering its dissociation properties, calcium chloride is not expected to

accumulate in living organisms.

Mobility in Soil: Chloride ions are mobile in soil eventually drainig into surface water.

13. Disposal Considerations

Waste Disposal Method: PRODUCT: Reuse or reprocess, if possible. Waste disposal must be done following all

Federal, State and Local regulations. Regulations may vary in different locations. Report spills if applicable. Chemical waste generators must determine whether a discarded chemical is classified as a hazardous waste. This product, if unaltered by use, may be disposed of by treatment at a permitted facility or as advised by your local waste

regulatory authority.

CONTAINER: Dispose properly accordingly to regulations on empty containers in your

locality or make available to a container reconditioning facility for recycling.

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14. Transport Information

LAND TRANSPORT (US DOT):

DOT Proper Shipping Name: Not Regulated. Trade Name: SETT ENHANCED

DOT Hazard Class: UN/NA Number:

MARINE TRANSPORT (IMDG/IMO):

IMDG/IMO Shipping Name: Not Regulated. Trade Name: SETT ENHANCED UN Number: Packing Group:

Hazard Class:

IMDG MFAG Number: N.A.

IMDG EMS Page:

AIR TRANSPORT (ICAO/IATA):

ICAO/IATA Shipping Name: Not Regulated. Trade Name: SETT ENHANCED

Additional Transport Placards / Markings: N.A.

Information:

Emergency Response Guide Number: N.A.

Reportable Quantity: N.A.

15. Regulatory Information

16. Other Information

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HEALTH 1
FLAMMABILITY 0
REACTIVITY 0
PPE

HMIS:



Additional Information About No data available.

This Product:

Company Policy or

Hazard Rating System:

Disclaimer: